

## Atomic Structure Guided Answers

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Name: \_\_\_\_\_ Period: \_\_\_\_\_ Unit 2 Atomic Structure- Guided Notes The Periodic Table \_\_\_\_\_ is credited for creating the first version of the modern periodic table \_\_\_\_\_ is a chart showing all the elements arranged in columns in such a way that all the elements in a given column exhibit similar chemical properties The periodic table is organized in order of \_\_\_\_\_ The periodic is organized in&mlr ...

~~Unit 2 Atomic Structure Guided Notes 2020.docx Name ...~~

The sum of the protons and neutrons in the nucleus of an atom.  $MASS\# = \# \text{ protons(atomic number)} + \# \text{neutrons}$ .  $\#NEUTRONS = Mass \# - \#Protons(atomic number)$   $\#Protons = Mass \# - \#Neutrons$ . ISOTOPES: Are atoms of the same element that have different numbers of. neutrons and different mass numbers.

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The average atomic mass of an element is usually closest to that of the isotope with the highest natural abundance. d. Because hydrogen has three isotopes with atomic masses of about 1 amu, 2 amu, and 3 amu, respectively, the average atomic mass of natural hydrogen is 2 amu. 19. Circle the letter of the correct answer.

~~5 ATOMIC STRUCTURE AND THE PERIODIC TABLE~~

~~Chapter 4 Atomic Structure Guided Practice Problems Answers Chapter 4 Atomic Structure. atom. atomic mass. atomic mass unit (amu) atomic number (Z) the smallest unit of an element that maintains the properties... the weighted average of the masses of the isotopes of an eleme... a unit of mass equal to one-twelfth the mass of a carbon-12 at... chapter 4 atomic structure Flashcards and Study Sets | Quizlet~~

~~Chapter 4 Atomic Structure Guided Practice Problems Answers~~

Unit 3 Atomic Structure- Guided Notes The Periodic Table ... - Add you answers from number 1 together - Round your answer to 2 decimal places • Practice: 1. A sample of Cesium is 75%  $^{133}\text{Cs}$ , 20%  $^{132}\text{Cs}$ , and 5%  $^{134}\text{Cs}$ . What is the average atomic mass? 2. Carbon-12 is 98% abundant and Carbon-14 is 2% abundant.

~~Name: Period: Unit 3 Atomic Structure Guided Notes~~

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made of metals. 2. The noble gases are the farthest group to the right on the Periodic Table. 3. Atoms having equal numbers of positive and negative charges make them neutral. 4. The three

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Protons and neutrons stick together to form the atomic nucleus. Electrons orbit around the nucleus. Chemical bonding and chemical reactions occur due to the electrons around atoms. An atom with too many or too few electrons is unstable and may bond with another atom to either share or essentially donate electrons.

~~Basic Atomic Structure and Atomic Theory — Study Guide~~

- Write the atomic number at the top of the square. - Write the element's name under the symbol. - Write the atomic mass at the bottom of the square. Step 2: Determine the number of protons, neutrons, and electrons in each element. Step 3: Create an atomic structure diagram for each element. Step 4: Draw the Bonding Structure for each element.

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Merely said, the atomic Atomic Structure Guided Notes Answers ATOMIC STRUCTURE DIAGRAMS Step 1: Complete the squares for each element by adding the atomic number, name and atomic mass. - Write the atomic number at the top of the square. - Write the element's name under the symbol. - Write the atomic mass at the bottom of the square.

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