

Chapter 9 Chemistry Test

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Download Chapter 9 Chemistry Test - Chapter 9 Practice Test Multiple Choice Identify the choice that best completes the statement or answers the question ****You need to know all the rules for naming ionic, molecular and acidic compounds ****You will have questions over ch 1,2,4 and 25 as well Keywords

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[4x10⁻¹⁰][3x4x10⁻¹⁰]³ -->6.9 x 10⁻³⁴ Q: What are the concentrations of each of the ions in a saturated solution of PbI2, given that the Ksp of PbI2 is 8.5 x10⁻⁹ at 25degC. If 5g pbI2 are dissolved in water to make 1L of solution at 25degC, would it be sat, unsat, or supersat

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